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Cobalt–zinc ferrite nano-particles $\text{Co}_{1-x}\text{Zn}_x\text{Fe}_2\text{O}_4$ were prepared by citrate precursor method, using cobalt, zinc, iron nitrates and citric acid as the organic precursor. Structural and morphological properties of the Ferrites were determined and characterized in detail by X-ray powder diffractometry (XRD), X-ray photo electron spectroscopy (XPS), Thermogravimetric analysis (TGA) and Infrared (IR) spectroscopy. X-ray diffraction pattern confirm the existence of single phase of cubic spinel crystal structure and the mean particle size of the nano-powders was calculated by Scherrer formula. Crystalline nature of the powder was observed when it was calcined at 500 °C. This study presents the effects of substitution of Zn^{2+} on the structural properties of cobalt-Zinc ferrite nano-particles. The observed variation in Crystalline size and increase in lattice constant are endorsed due to the Zn^{2+} substitution in ferrites is reported. The unit cell parameter is found to increases linearly with increasing concentration of zinc due to larger ionic radii of Zn^{2+} ion. Fourier transformed infrared (FT-IR) spectra of Zn^{2+} ions substituted in Co-Zn spinal ferrite nano-particles have been analyzed in frequency range of 4000-400 cm^{-1} . Two absorption bonds in IR spectra were observed in the frequency range of 600-400 cm^{-1} may be caused by the metal-oxygen vibrations in octahedral and tetrahedral sites.

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